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## Structural Analysis of GaInN Single Quantum Well using X-ray Absorption Fine Structure

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Although Ga<sub>1-x</sub>In<sub>x</sub>N quantum wells have been used for the emitting layer of a GaN-based laser diode, which is a key device in high-density optical storage systems, there is a serious problem in that phase separation is occasionally observed in Ga<sub>1-x</sub>In<sub>x</sub>N alloy with a high indium content (x>0.20). This separation can be ascribed to the large lattice-constant difference between GaN and InN, but the details are not known. We have already analyzed the local structures around In atoms of Ga<sub>1-x</sub>In<sub>x</sub>N layer [1] and Ga<sub>1-x</sub>In<sub>x</sub>N /GaN multi-quantum wells [2] using the fluorescence-detecting x-ray absorption fine structure (XAFS) method. Here we investigated Ga<sub>1-x</sub>In<sub>x</sub>N/GaN single quantum wells, which is useful for the above discussion although it is difficult to obtain the XAFS signal because of the thin layer.

As a sample, a 2 $\mu$ m-thick GaN layer was grown on a sapphire substrate using metal-organic chemical vapor deposition (MOCVD), followed by a single quantum well of Ga<sub>0.92</sub>In<sub>0.08</sub>N (4.5nm thick). Fluorescence-detecting XAFS experiments were performed at BL01B1 of SPring-8 using synchrotron radiation from the 8 GeV storage ring. The x-ray was monochromated by two Si (111) crystals and irradiated on the sample. Fluorescence x-ray was obtained from the sample and detected by a Ge solid-state multi-detector with 19 elements.

Figure 1 shows the In K-edge

fluorescence-vield spectrum for Ga<sub>0.92</sub>In<sub>0.08</sub>N/GaN single quantum well measured at 300K. The intensity of fluorescence x-ray was corrected in order to reflect the solid-state detector's inability to accurately detect high-intensity x-ray. The absorption K-edge of In atoms was observed near the excited photon energy of 27.94 keV. A clear fine structure was also observed at an energy above the In K-edge. We confirmed that the local structures around In atoms of a Ga<sub>1-x</sub>In x N/GaN could be analyzed using XAFS system of SPring-8.

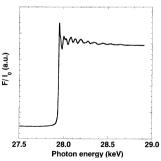


Fig.1 In K-edge fluorescence-yield spectrum of Ga<sub>0.92</sub>In<sub>0.08</sub>N/GaN single quantum well

## References

- [1] T. Miyajima, et al., SPring-8 User Experiment Report No.5 (2000A) 393.
- [2] T. Miyajima, et al., SPring-8 User Experiment Report No.6 (2000B) 2.

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## Investigation of local structure of Re active site of Re/ $\gamma$ -Fe<sub>2</sub>O<sub>3</sub> catalysts by Re-K edge EXAFS

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Rhenium oxide supported on  $Fe_2O_3$  has a unique property in the methanol selective oxidation reaction to produce methylal selectively. We measured Re K–edge EXAFS at BL01B1 to elucidate the local structure of rhenium species in  $ReO_x/\gamma$ - $Fe_2O_3$  and  $ReO_x/Al_2O_3$ .

The spectra were recorded at 8 K in a transmission mode. A Si(511) double crystal was used as monochromator and the X-ray intensities before and after sample were monitored by ionization chambers filled with pure Ar and Xe, respectively. The spectra were analyzed by the UWX-AFS package. The phase shift and backscattering amplitude were calculated by the FEFF8 code and the coefficient of the multi-photon effects of rhenium cations was calculated from the EXAFS of rhenium oxides.

Figure 1 shows Fourier transformed Re K–edge EXAFS functions  $(k^3\chi(k))$  of fresh  ${\rm ReO}_x/\gamma{\rm -Fe_2O_3}$  and  ${\rm ReO}_x/{\rm Al_2O_3}$  at various Re loadings. Phase shifts are not corrected in the figure

In the spectra for low loaded ReO<sub>r</sub>/ $\gamma$ -Fe<sub>2</sub>O<sub>3</sub> (1.7 and 3.4 wt%), only Re-O contribution was observed at 0.175 nm and the contributions for longer distances such as Re-(O)-Re and Re-(O)-Fe were not observed. The Re-O distance and coordination number (CN) were not affected by the loading and were 0.175 nm and 4 in 1.7– 9.2 wt%. This result together with the result of Re L<sub>I</sub> edge XANES suggests that Re species on  $\gamma$ -Fe<sub>2</sub>O<sub>3</sub> has tetrahedral ReO<sub>4</sub> structure regardless the Re loading. For ReO<sub> $\tau$ </sub>/ $\gamma$ -Fe<sub>2</sub>O<sub>3</sub> (9.2) wt%), a Re-(O)-Fe contribution was observed at 0.384 nm with the CN of 1.0. The longer distance contribution cannot be fitted by a Re-O-Re contribution. This result together with the XRD result suggests that crystalline Re-Fe binary oxide was formed in  $ReO_x/\gamma$ -Fe<sub>2</sub>O<sub>3</sub> (9.2 wt%). As the Re-O contribution was not affected by the loading, similar species to those at 9.2 wt% were expected to be formed at 1.7 and 3.4 wt%. The absence of the longer distance contributions at low loaded samples can be explained by a large statistical disorder.

For ReO<sub>x</sub>/Al<sub>2</sub>O<sub>3</sub> catalysts, on the contrary, Re–O contributions were affected by loading. at 1.0 wt% only Re=O was observed at 0.173 nm with CN being 2.9. A Re–O contribution was observed at 0.200 nm for ReO<sub>x</sub>/Al<sub>2</sub>O<sub>3</sub> 2.0 wt% sample and the ratio of Re–O increased with increasing Re loading. In addition, longer distance contributions, which can be assigned to Re–(O)–Re contributions of ReO<sub>2</sub> were observed at 6.0 wt%. Thus, in the case of ReO<sub>x</sub>/Al<sub>2</sub>O<sub>3</sub>, the local structure around Re is affected by loading. At low loading, Re exist as isolated species and at high loading Re exist as ReO<sub>2</sub>.

We found that Re–Fe binary oxide was formed in  $\text{ReO}_x/\gamma$ -Fe<sub>2</sub>O<sub>3</sub> catalysts. Such a binary oxide did not formed on  $\text{ReO}_x/\text{Al}_2\text{O}_3$  catalysts. The formation of Re–Fe binary oxide can be a reason for high performance and long lifetime of  $\text{ReO}_x/\gamma$ -Fe<sub>2</sub>O<sub>3</sub> catalysts.

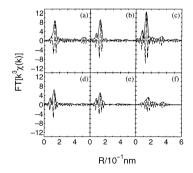


Figure 1. Fourier transformed Re–K edge EXAFS functions  $(k^3\chi(k))$  of fresh  ${\rm ReO}_x/\gamma$ - ${\rm Fe_2O_3}$  (1.7 wt%) (a),  ${\rm ReO}_x/\gamma$ - ${\rm Fe_2O_3}$  (3.4 wt%) (b),  ${\rm ReO}_x/\gamma$ - ${\rm Fe_2O_3}$  (9.2 wt%) (c),  ${\rm ReO}_x/{\rm Al_2O_3}$  (1.0 wt%) (d),  ${\rm ReO}_x/{\rm Al_2O_3}$  (2.0 wt%) (e), and  ${\rm ReO}_x/{\rm Al_2O_3}$  (9.2 wt%) (f).